



(19)

Europäisches Patentamt

European Patent Office

Office européen des brevets



(11)

EP 1 227 143 A1

(12)

## EUROPEAN PATENT APPLICATION

(43) Date of publication:  
31.07.2002 Bulletin 2002/31(51) Int Cl.<sup>7</sup>: C10L 1/14, C10L 1/10,  
C10L 1/02, C10L 1/32,  
C10L 10/02

(21) Application number: 01101960.1

(22) Date of filing: 29.01.2001

(84) Designated Contracting States:  
AT BE CH CY DE DK ES FI FR GB GR IE IT LI LU  
MC NL PT SE TR  
Designated Extension States:  
AL LT LV MK RO SI

- Cognis Deutschland GmbH & Co. KG  
40589 Düsseldorf (DE)

(71) Applicants:  
• AAE Technologies International Limited  
Dublin 2 (IE)

(72) Inventor: The designation of the inventor has not  
yet been filed

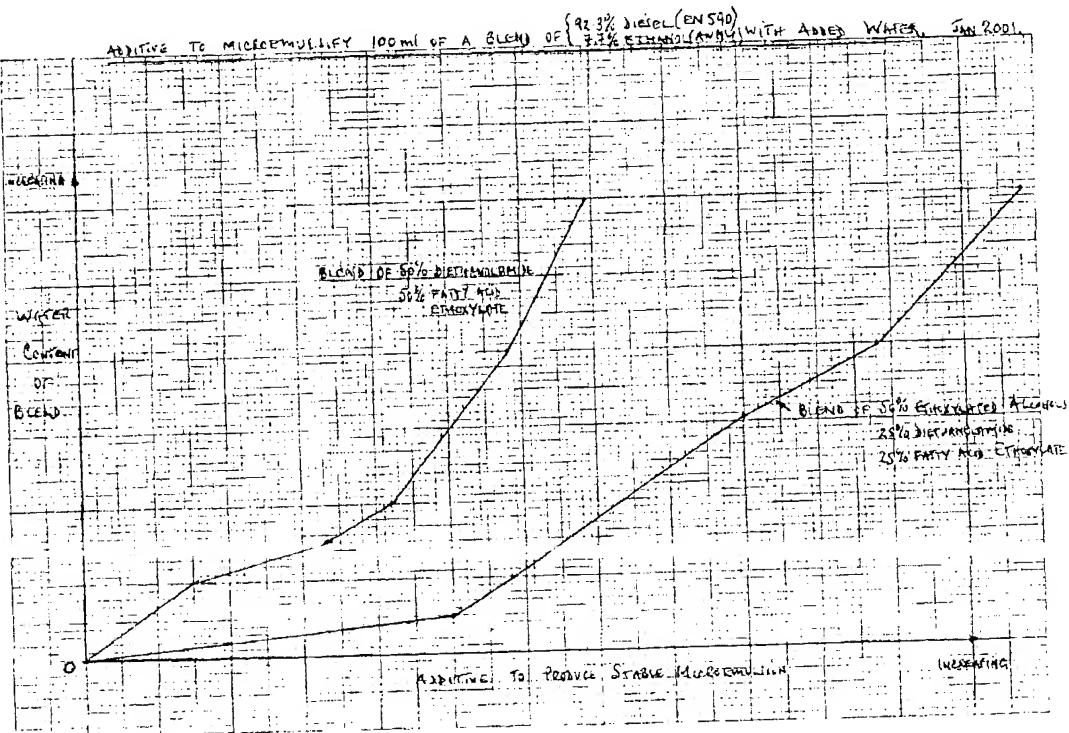
(74) Representative: Harrison Goddard Foote  
Belgrave Hall  
Belgrave Street  
Leeds LS2 8DD (GB)

## (54) Fuel additives

(57) There is described a fuel additive composition comprising an oleic alkanolamide and an alkoxylated oleic acid.

There is also described a fuel composition compris-

ing a hydrocarbon fuel and a fuel additive comprising an oleic alkanolamide and an alkoxylated oleic acid and also a method of running a combustion engine using the fuel if the invention.



EP 1 227 143 A1

**Description**

[0001] This invention relates to a novel fuel additive, to methods of their preparation and use and to novel fuel formulations.

5 [0002] Surfactants have long had utility as additives which can affect the performance of fuels such as gasoline and diesel. British Patent No 2217229 describes a solubilising compound as a fuel additive. In particular there is described a composition comprising:

10 48 parts by volume of an alcohol ethoxylate;  
 3-8 parts by volume of lauric diethanolamide;  
 3-8 parts by volume of oleic diethanolamide; and  
 1.5-4 parts by volume of a polyglycol ether such as ethoxylated oleic acid.

15 [0003] Such compositions are useful as fuel additives and enable the solubilisation of water in fuels thus reducing its potential corrosive effect. However, the compositions are disadvantageous in that, *inter alia*, they require a high additive to fuel ratio. Furthermore, they do not address the problems of emissions of gases such as CO, CO<sub>2</sub> and NO<sub>x</sub>.  
 [0004] International Patent Application No WO 98/17745 describes an alternative formulation which comprises,

20 25% w/w of a diethanolamide,  
 50% w/w of an ethoxylated alcohol, and  
 25% w/w of a fourteen carbon chain fatty acid with seven ethoxylate groups.

25 [0005] WO '745 especially describes fuel compositions comprising, *inter alia*, an additive made up of a fatty acid diethanolamide, an alcohol ethoxylate and an ethoxylate of a fatty acid, the degree of ethoxylation being selected so that a long term stable fuel composition is formed and, in particular, wherein, by carefully selecting the degree of ethoxylation, a balanced blend can be produced.

30 [0006] Whilst such additives provide significant reductions in emissions and are useable at low concentrations, there is still a need for a fuel composition which is capable of reducing emissions whilst maintaining performance levels and which may be even more cost effective to use.

[0007] We have now surprisingly found that a more advantageous additive can be prepared without the use of an ethoxylated alcohol

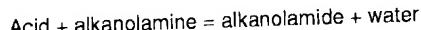
35 [0008] Thus according to the invention we provide a fuel additive composition comprising an oleic alkanolamide and an alkoxylated oleic.

[0009] A fuel additive composition as hereinbefore described is especially characterised in that alcohols, and especially ethoxylated alcohols, are substantially absent from the composition. Therefore we provide fuel additive as hereinbefore described which is substantially free of alcohol, especially ethoxylated alcohol.

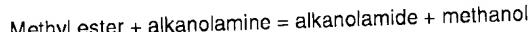
40 [0010] The oleic alkanolamide is preferably an ethanolamide and more preferably a diethanolamide. Especially preferred are the diethanolamides and particularly the super diethanolamides.

[0011] Generally, there are three commercial routes to alkanolamides;

45



45



50 [0012] These are listed in order of increasing product quality. The route via the acid often uses an excess of alkanolamine to produce a product higher in amide than is obtainable from the acid. A stoichiometric ratio is used; these products are sometimes referred to as super amides.

55 [0013] The oleic acid ethoxylate may be derived from a variety of feedstocks, readily available worldwide. However, in a preferred embodiment of the invention the oleic acid ethoxylate may be produced by ethoxylation or esterification of acids derived from animal fats e.g. beef tallow or vegetable oils, such as soya, etc. Thus the oleic acid precursor may be predominantly, e.g. from 65-70% v/v, oleic acid, but may also contain linoleic acid, e.g., 10-12% v/v, and may also include small amounts of stearic, palmitic and/or myristic acids.

[0014] The alkoxylate may be an ethoxylate or a propyoxylate or a mixture thereof. The degree of ethoxylation is

chosen to optimise performance in the blend with the oleic diethanolamide and may be from 0.5 to 20, but more preferably from 0.5 to 10, more preferably from 1 to 3. A suitable product within this range would be, for example, that derived from the addition of 1 molecule of ethylene oxide to 1 mole of oleic acid.

5 [0015] The ratio of the oleic acid alkoxylate to the oleic alkanolamide may vary, but may be from 99:1 to 1:99 v/v, preferably 3:1 to 1:1 v/v and most preferably 1:1v/v.

[0016] The additive of the invention may be added to any known hydrocarbon fuel, e.g. diesel, petrol or alcohol, such as ethanol, which may or may not contain water. The invention is seen to particularly good effect when added to fuels based on low fraction oils.

10 [0017] Thus according to a further feature of the invention we provide a fuel composition comprising a hydrocarbon fuel and a fuel additive as hereinbefore described.

[0018] The concentration of the additive in the fuel composition may vary depending, *inter alia*, upon the nature of the fuel, however, the concentration can be very low, typically of the order of from 0.5:1000 to 50:1000 v/v, preferably from 1:1000 to 50:1000 v/v, preferably 1:100 to 5:100 v/v. The hydrocarbon fuel may comprise any known hydrocarbon fuel or mixtures thereof, therefore such fuels include but shall not be limited to diesel, e.g., petroleum diesel or biodiesel, gasoline, aviation fuel, alcohol, etc.

15 [0019] In one embodiment of the fuel composition of the invention the hydrocarbon fuel is a petroleum diesel fuel. Such fuels may generally be obtained from the distillation of petroleum and its efficiency can be measured by the cetane number. Suitable diesel fuels for use in accordance with the invention generally have a cetane number of from 35 to 60, preferably from 40 to 50. The amount of diesel fuel blended to form the fuel composition of the invention may be from 60 % v/v to 95 % v/v, based on the total volume of the fuel consumption. The diesel may comprise petroleum diesel, biodiesel or biodiesel blended with petroleum diesel in any ratio of from 99:1 to 1:99 v/v.

20 [0020] Preferably, when the fuel of the invention is blend of biodiesel and a petroleum diesel it may comprise up to 50% v/v biodiesel, for example from 1 to 50% v/v, preferably from 5 to 30% v/v, more preferably from 10 to 20% v/v.

[0021] In a further feature of the invention the hydrocarbon fuel, such a diesel or gasoline may include an amount of an oxygenator, e.g. alcohol, an alkanol, such as ethanol. When an alcohol is present the amount of alcohol may vary depending, *inter alia*, upon the nature of the fuel, but may in an amount of from 1 to 50% v/v, preferably 5 to 20% v/v.

25 [0022] For fuels, ethanol may be produced from fossil fuel feedstocks or by fermentation of sugars derived from grains or other biomass materials. Therefore, ethanol suitable for use in accordance with the fuel compositions of the invention may be fuel grade ethanol derived from yeast or bacterial fermentation of starch-based sugars. Such starch-based sugars may be extracted from corn, sugarcane, tapioca and sugar beet.

30 [0023] Alternatively, fuel grade ethanol may be produced via known dilute and/or concentrated acid and/or enzymatic hydrolysis of a particular biomass material, for example, from waste industrial sources including, cellulosic portions of municipal solid waste, waste paper, paper sludge, saw dust. Biomass may also be collected from agricultural residues including, for example, rice husks and paper-mill sludge.

35 [0024] A suitable fuel grade ethanol for use in accordance with the invention may contain none or only contaminant levels of water. Alternatively, a suitable fuel grade ethanol for use in accordance with the invention may contain higher amounts of water, up to 5% w/w (hydrous ethanol).

40 [0025] Use of ethanol in combination with a diesel fuel has previously posed problems wherein the ethanol/diesel resultant mixture would undesirably separate into two distinct phases, especially when water is present, and render the fuel mixture unsuitable for use as a combustible fuel. The use of the fuel additives of the invention permits hydrous ethanol to be blended satisfactorily with conventional diesel fuel without forming two phases. The use of fuel grade ethanol blended in accordance with the invention imparts desirable combustion characteristics to the overall fuel composition; such as improved fuel stability, lower smoke and particulate matter, lower CO and NO<sub>x</sub> emissions, improved antiknock characteristics, and/or improved anti-freeze characteristics.

45 [0026] When the fuel compositions of the invention are described, the absence of alcohol from the fuel additive compositions should not be construed as meaning that alcohol, e.g. ethanol, is absent from the fuel composition.

[0027] The presence of the additive of the invention ensures that the fuel composition forms a consistent stable homogenous composition and creates a monolayer simultaneously a result of which leads to a better more complete burn which reduces pollution and increases miles per gallon.

50 [0028] As a result a blended fuel, particularly alcohol based, is able to combust more precisely with a cooler charge to reduce the iron-formates present from the aldehyne peracids and peroxide reactions normally attributable to engine degradation.

[0029] In a further aspect of the invention we provide a fuel composition comprising a liquid hydrocarbon fuel and a surfactant composition as hereinbefore described.

55 [0030] In the fuel composition of the invention the hydrocarbon fuel, may be any conventionally known fuel, e.g. gasoline, diesel, biodiesel, etc. Furthermore the fuel of the invention may, optionally, include an oxygenator. The oxygenator may be an alcohol, such as ethanol. When an oxygenator such as ethanol is present, the amount of ethanol in the fuel may vary and may be from 0 to 25% v/v ethanol, preferably from 1 to 10% v/v ethanol.

[0031] We further provide a method of running an internal combustion engine comprising the use of a fuel composition of the invention.

[0032] International Patent Application No. WO99/35215, Wenzel, describes an additive for combustible fuels which includes a nitrogen source, such as urea. Whilst the additive is said to reduce NOx, the compositions are very complex and include numerous ingredients, including:

- 5 a water soluble alcohol,
- 10 a C6 to C12 alcohol,
- a C6 to C18 ethoxylated alcohol,
- 15 a C10 to C24 fatty acid, and
- a nitrogen source.

[0033] We have now surprisingly found that the aforementioned additives are suitable for use in very low fuel: additive ratios in combination with nitrogenous compounds such as urea.

[0034] Thus according to the invention we provide a fuel additive composition comprising an oleic alkanolamide, an alkoxylated oleic acid as hereinbefore described and a nitrogen compound.

[0035] The nitrogen compound may be selected from the group consisting of ammonia, hydrazine, alkyl hydrazine, dialkyl hydrazine, urea, ethanolamine, monoalkyl ethanolamine, and dialkyl ethanolamine wherein alkyl is independently selected from methyl, ethyl, n-propyl or isopropyl. Urea is preferred. The nitrogen compound may be an anhydrous compound or a hydrous compound, e.g. an aqueous solution, and may be up to a 5% w/w aqueous solution.

[0036] According to a yet further feature of the invention we provide a method of solubilising a nitrogen compound in a fuel composition which comprises mixing a hydrocarbon fuel, a nitrogen compound and a fuel additive as hereinbefore described. The method of the invention may optionally include the addition of an alcohol, such as ethanol or water, as hereinbefore described.

[0037] We also provide the use of a nitrogen compound in the manufacture of a fuel additive of the invention. We especially provide the use of urea in the manufacture of a fuel additive of the invention.

[0038] We further provide a fuel composition comprising a hydrocarbon fuel, a fuel additive as hereinbefore described and a nitrogen compound.

[0039] In the fuel composition in this aspect of the invention the nitrogen compound may be added by being incorporated into the fuel additive or may be added separately. Furthermore, the nitrogen compound may be added as an aqueous solution.

[0040] A particular advantage of the present invention over the prior art is that fuel compositions may be prepared which are substantially anhydrous, save for trace water contamination. By the term trace water contamination we generally mean 0.1 % w/w water or less.

[0041] Thus, according to a yet further feature of the invention we provide the use of urea in the manufacture of a fuel composition as hereinbefore described.

[0042] The fuel additive or the fuel composition of the invention may also optionally comprise a cetane booster in the amount of from 0.1 % v/v to 10 % v/v, based on the volume of the mixture. When a cetane booster is included in the fuel composition of the invention it may be added as part of the fuel additive of the invention or it may be added separately.

[0043] A suitable cetane booster for use in the mixture is selected from the group comprising, 2-ethylhexyl nitrate, tertiary butyl peroxide, diethylene glycol methyl ether, cyclohexanol, and mixtures thereof. The amount of cetane booster present in the mixture is a function of the cetane value of the particular diesel fuel and the amount of ethanol present in the particular fuel composition. Generally, the lower the diesel fuel cetane value, the higher the amount of the cetane booster. Similarly, because ethanol typically acts as a cetane depressant, the higher the concentration of ethanol in the solution, the more cetane booster may be necessary in the mixture.

[0044] The fuel additives of the invention are advantageous in that, inter alia, they are more efficient at producing micro emulsions than prior art additives. Therefore, they are capable of more efficiently producing a stable, clear and homogeneous solution with a hydrocarbon fuel, e.g. diesel/ethanol, even in the presence of water. Therefore, according to a further feature of the invention we provide a fuel composition as hereinbefore described, which optionally includes an amount of water, and wherein the fuel consists of a substantially stable, substantially clear and substantially homogeneous solution.

[0045] Furthermore, the fuel additive or the fuel composition of the invention may also optionally include a demulsifier in an amount of less than 5 % v/v and preferably less than 1 % v/v based on the volume of the mixture.

[0046] A measure of when a fuel composition is at or near its cloud point is the conductivity of the fuel. For example, water has a conductivity of 100 mS cm<sup>-1</sup> and an alcohol, e.g. ethanol, a conductivity of 20 to 30 mS cm<sup>-1</sup>. Fuels, such as gasoline or diesel, being non-polar, have a conductivity of substantially zero. Furthermore, we have found that a non-homogenous mixture of a fuel, such as gasoline or diesel, optionally including an alcohol, such as ethanol, as

hereinbefore described, will have a relatively high conductivity reading, and as homogeneity is approached, the conductivity will reduce and will reach a minimum when the composition is a clear homogenous solution.

[0047] Thus according to a further feature of the invention we provide a method of determining the homogeneity of a fuel composition which comprises measuring the conductivity of the composition.

[0048] The conductivity may be measured at varying temperatures. However, we have found that measuring at substantially ambient temperature is preferred and particularly at 25.1°C. Conductivity values given hereinafter generally relate to such values when measured at 25.1°C. Furthermore, since it is known that conductivity may vary with temperature, any conductivity values taken at differing temperatures should be calibrated as if measured at 25.1°C.

[0049] We especially provide a method of determining the homogeneity of a fuel composition which comprises a fuel and an oxygenator.

[0050] The preferred additive of this invention is a non-ionic surfactant and preferably a blend of surfactants. It is a preferred feature of this invention that the surfactants be selected by their nature and concentration that, in use, the additive (as well as any water or other non-fuel liquid present) be solubilised within the fuel. For this purpose it is convenient to have regard to the hydrophilic-lipophilic (HLB) of the surfactant, the value being calculated according to the expression.

$$HLB = \frac{\text{mol. wt of hydrophilic chain} \times 20}{\text{total mol. wt.}}$$

[0051] The values will depend on the length of the hydrophilic chain, typically an ethoxylate chain. The length of the chain will increase the extent of solubilisation because of a greater ability to solubilise.

[0052] The invention has the ability to unify the HLB requirements of any liquid fuel which in turn allows for one dose to be used in any fuel from C5 carbon chains up. The benefit being the amount of treatment directly related to the co-solvency ability.

[0053] The invention will now be illustrated, but in no way limited, with reference to the accompanying examples.

### Example 1

#### Preparation of Compositions

[0054] An additive composition was made up by blending constituents, the super diethanolamide of oleic acid ethoxylated oleic acid in the ration 1:1. 1% of this composition was added to 7.7% ethanol/ 92.3% diesel blends, including certification diesel, US No 1 diesel, 10% aromatic diesel containing 0.1% cetane improver resulting in optically clear and stable micro-emulsion fuels. These were tested as automotive fuels on a 1991 Detroit Diesel Series 60 engine using the EPA (USA Environmental Protection Agency) heavy duty engine certification test as described in the Code of Federal Regulations, Title 40, Part 86, Subpart N.

[0055] Toxic exhaust gas emissions were measured and compared with those of the base unadditised diesels. Significant reductions were obtained of toxic gases CO, CO<sub>2</sub>, NOx and particulate matter and the results are shown in Tables I - III.

### Example 2

#### Test Protocol

[0056] Blends were made up of diesel, anhydrous ethanol and water added in concentrations up to 5% v/v. The cloudy blends were then titrated against the additive until clear, stable, micro-emulsions were formed. The volume of additive required to micro-emulsify the diesel/ethanol/water blend was plotted against water content and is shown in Figure 1. The relative efficiency of additives can be compared using this test protocol.

[0057] Figure 1 shows a comparison of the performance of an additive comprising oleic diethanolamide and an ethoxylated oleic acid in a 1:1 ratio compared with an additive of the prior art comprising a fatty acid ethoxylate a diethanolamide and an ethoxylated alconol in a ratio of 1:1.2.

### Example 3

#### Petroleum Diesel/Biodiesel/Ethanol Blend

[0058] A blend was made of certification diesel, biodiesel and ethanol, stabilised with the surfactant additive described in Example 1. Emission tests showed reductions in CO, NOx and particulate matter compared with base diesel.

**Example 4**

**Petroleum Diesel/Ethanol/Urea Blend**

5 [0059] Example A solution of 0.25% urea in ethanol was blended in a 7.7.92.3 ratio with US No 1 diesel and 1.0% of the additive described in Example 1, was added to produce a clear micro-emulsion automotive fuel. Tests using this blend showed that toxic gas emissions were again lower than the base fuel, the urea having made a contribution to the reductions obtained.

10 **Example 5**

**Gasoline/ethanol blends**

15 [0060] Blends were made up of EPA and CARB gasolines with various amounts of ethanol e.g. a typical blend contained 90% v/v gasoline and 10% ethanol. Small amounts, typically 1% v/v of the additive of the invention were added with those of the base gasolines. The results confirmed that ethanol/gasoline blends combust to give lower levels of toxic gas emissions.

20

25

30

35

40

45

50

55

Test protocol and results to be supplied.

[0061]

5

10

15

20

25

30

35

40

45

50

55

Table I

No. 1 Diesel Comparisons		Map	BHP-hr	g/BHP-hr	-hr	THC	NOx	CO	CO <sub>2</sub>	PM
FUEL	Deviance									
Cert Fuel			21.256	0.090	4.512	3.849	562.56	0.196		
Philips No 1			<b>0.000</b>							
Philips No 2 ( for			21.733	0.057	4.648	5.013	567.61	0.262		
comparison)										
<i>Deviance from Cert</i>										
Cert + 7.7% AAE blend +										
2000 EHN			20.940	0.163	4.425	3.092	558.74	0.142		
Deviance from Cert										

5  
10  
15  
20  
25  
30  
35  
40  
45  
50  
55

Table II

10% Aromatic Diesel (CARB Equivalent) Comparisons						
FUEL	Map BHP-hr	THC g/BHP-hr	NOx g/BHP-hr	CO g/BHP-hr	CO <sub>2</sub> g/BHP-hr	PM g/BHP-hr
Cert Fuel Philips No.2 <i>Deviance from Cert.</i>	21.733 <i>0.000</i>	0.057 <i>0.000</i>	4.648 <i>0.000</i>	5.013 <i>0.000</i>	567.61 <i>0.000</i>	0.262 <i>0.000</i>
Cert +7.7% AAE blend + 1000 EHN <i>Deviance from Cert.</i>	21.538 <i>-0.90%</i>	0.094 <i>64.91%</i>	4.668 <i>0.43%</i>	3.777 <i>-24.66%</i>	565.00 <i>-0.46%</i>	0.174 <i>-33.59%</i>
Cert +8.7% AAE blend +11.3% Bio + 1000 EHN <i>Deviance from Cert.</i>	21.350 <i>-1.76%</i>	0.098 <i>71.93%</i>	4.630 <i>-0.39%</i>	3.789 <i>-24.42%</i>	565.67 <i>-0.34%</i>	0.161 <i>-38.55%</i>

5

10

15

20

25

30

35

40

45

50

**Claims**

1. A fuel additive composition comprising an oleic alkanolamide and an alkoxylated oleic acid.
2. A fuel additive composition according to claim 1 **characterised in that** the additive is substantially free of ethoxylated alcohol.
3. A fuel additive composition according to claim 1 **characterised in that** the oleic alkanolamide is an ethanolamide.

Table III

<b>10% Aromatic Diesel (CARB Equivalent) Comparisons</b>						
<b>FUEL</b>	<b>Map</b>	<b>THC</b> <b>BHP-hr</b>	<b>NO<sub>x</sub></b> <b>g/BHP-hr</b>	<b>CO</b> <b>g/BHP-hr</b>	<b>CO<sub>2</sub></b> <b>g/BHP-hr</b>	<b>PM</b> <b>hr</b>
Cert Fuel						
Philips 10% Aromatic	21.757	0.050	4.318	4.714	554.78	0.249
<i>Deviance</i>	<b>0.000</b>	<b>0.000</b>	<b>0.000</b>	<b>0.000</b>	<b>0.000</b>	<b>0.000</b>
Cert plus AAE05 only	21.735	0.041	4.332	4.885	556.88	0.245
<i>Deviance from Cert</i>	<b>-0.10%</b>	<b>-18.00%</b>	<b>0.32%</b>	<b>3.63%</b>	<b>0.38%</b>	<b>-1.61%</b>
Cert +7.7% AAE blend +	21.402	0.122	4.365	3.527	556.14	0.150
1000 EHN						
<i>Deviance from Cert</i>	<b>-1.63%</b>	<b>144.00%</b>	<b>1.09%</b>	<b>-25.18%</b>	<b>0.25%</b>	<b>-39.76%</b>
Cert +7.7% AAE blend +	21.272	0.116	4.344	3.670	559.34	0.161
2000DTBP						
<i>Deviance from Cert</i>	<b>-2.23%</b>	<b>532.00%</b>	<b>0.60%</b>	<b>-22.15%</b>	<b>0.82%</b>	<b>-35.34%</b>

4. A fuel additive composition according to claim 3 **characterised in that** the oleic ethanolamide is a diethanolamide.
5. A fuel additive composition according to claim 1 **characterised in that** the alkoxylated oleic acid is an ethoxylated oleic acid, propuloxylated oleic acid or a mixture thereof.
6. A fuel additive according to claim 5 **characterised in that** the alkoxylated oleic acid is an ethoxylated oleic acid.
7. A fuel additive composition according to claim 5 **characterised in that** the degree of alkoxylation is from 0.5 to 10 mol of alkoxylate to 1 mol of oleic acid.
8. A fuel additive composition according to claim 7 **characterised in that** the degree of alkoxylation is 1 mol of alkoxylate to 1 mol of oleic acid.
9. A fuel additive composition according to claim 1 **characterised in that** the precursor of the alkoxylated oleic acid comprises 65-70% v/v oleic acid.
10. A fuel additive composition according to claim 9 **characterised in that** the precursor of the alkoxylated oleic acid comprises other acids selected from one or more of linoleic acid, stearic acid, palmitic acid and myristic acid.
11. A fuel additive composition according to claim 1 **characterised in that** the ratio of oleic diethanolamide to ethoxylated oleic acid is from 99:1 to 1:99 v/v.
12. A fuel additive according to claim 11 **characterised in that** the ratio of oleic diethanolamide to ethoxylated oleic acid is 1:1.
13. A fuel composition comprising a hydrocarbon fuel and a fuel additive according to claim 1.
14. A fuel composition according to claim 13 **characterised in that** the hydrocarbon fuel is selected from one or more of a diesel, e.g., petroleum diesel or biodiesel, gasoline, aviation fuel and an alcohol.
15. A fuel composition according to claim 13 **characterised in that** composition includes an amount of water and the fuel consists of a substantially stable, substantially clear and substantially homogeneous solution.
16. A fuel composition according to claim 13 **characterised in that** the additive to fuel ratio is from 0.5 - 50:1000 v/v.
17. A fuel composition according to claim 16 **characterised in that** the additive to fuel ratio is from 1:1000 to 50:1000 v/v
18. A fuel composition according to claim 17 **characterised in that** the additive to fuel ratio is from 1 to 5:100 v/v.
19. A fuel composition according to claim 13 **characterised in that** the fuel is a diesel fuel.
20. A fuel composition according to claim 19 **characterised in that** the diesel fuel is a petroleum diesel.
21. A fuel composition according to claim 20 **characterised in that** the diesel fuel is a blend of petroleum diesel and biodiesel
22. A fuel composition according to claim 20 **characterised in that** the diesel fuel is a biodiesel.
23. A fuel composition according to claim 19 **characterised in that** the fuel is a mixture of diesel and an alcohol.
24. A fuel composition according to claim 23 **characterised in that** the alcohol is ethanol.
25. A fuel composition according to claim 23 **characterised in that** the fuel is a hydrous ethanol/ diesel blend.
26. A fuel composition according to claim 25 **characterised in that** the additive to fuel ratio is up to 5% v/v.
27. A fuel composition according to claim 23 **characterised in that** the fuel is an anhydrous ethanol/diesel blend.

28. A fuel composition according to claim 13 **characterised in that** the fuel is gasoline.

29. A fuel composition according to claim 28 **characterised in that** the fuel is gasoline/ethanol blend.

5 30. A fuel composition according to claim 29 **characterised in that** the additive to fuel ratio is up to 5% v/v.

31. A fuel composition according to claim 13 **characterised in that** a nitrogen compound is also present.

10 32. A fuel composition according to Claim 31 **characterised in that** the nitrogen compound is selected from the group consisting of the ammonia, hydrazine, alkyl hydrazine, dialkyl hydrazine, urea, ethanolamine, monoalkyl ethanolamine, and dialkyl ethanolamine wherein alkyl is independently selected from methyl, ethyl, n-propyl or isopropyl.

33. A fuel composition according to claim 32 **characterised in that** the nitrogen compound is urea.

15 34. A method of solubilising a nitrogen compound in a fuel composition which comprises mixing a hydrocarbon fuel, a nitrogen compound and a fuel additive composition according to claim 1.

35. The use of a nitrogen compound in the manufacture of a fuel composition according to Claim 31.

20 36. The use according to claim 35 **characterised in that** the nitrogen compound is urea.

37. A fuel additive according to claim 1 **characterised in that** the additive also comprises a cetane booster in amount of from 0.1 % v/v to 10 % v/v.

25 38. A fuel composition according to claim 13 **characterised in that** the fuel composition also comprises a cetane booster in amount of from 0.1 % v/v to 10 % v/v.

39. A fuel additive according to claim 1 **characterised in that** the additive also comprises a demulsifier in an amount of less than 5 % v/v.

30 40. A fuel composition according to claim 13 **characterised in that** the fuel composition also comprises a demulsifier in an amount of less than 5 % v/v.

35 41. The use of an oleic alkanolamide in the manufacture of a fuel additive according to claim 1.

42. The use of an alkoxylated oleic acid in the manufacture of a fuel additive composition according to claim 1.

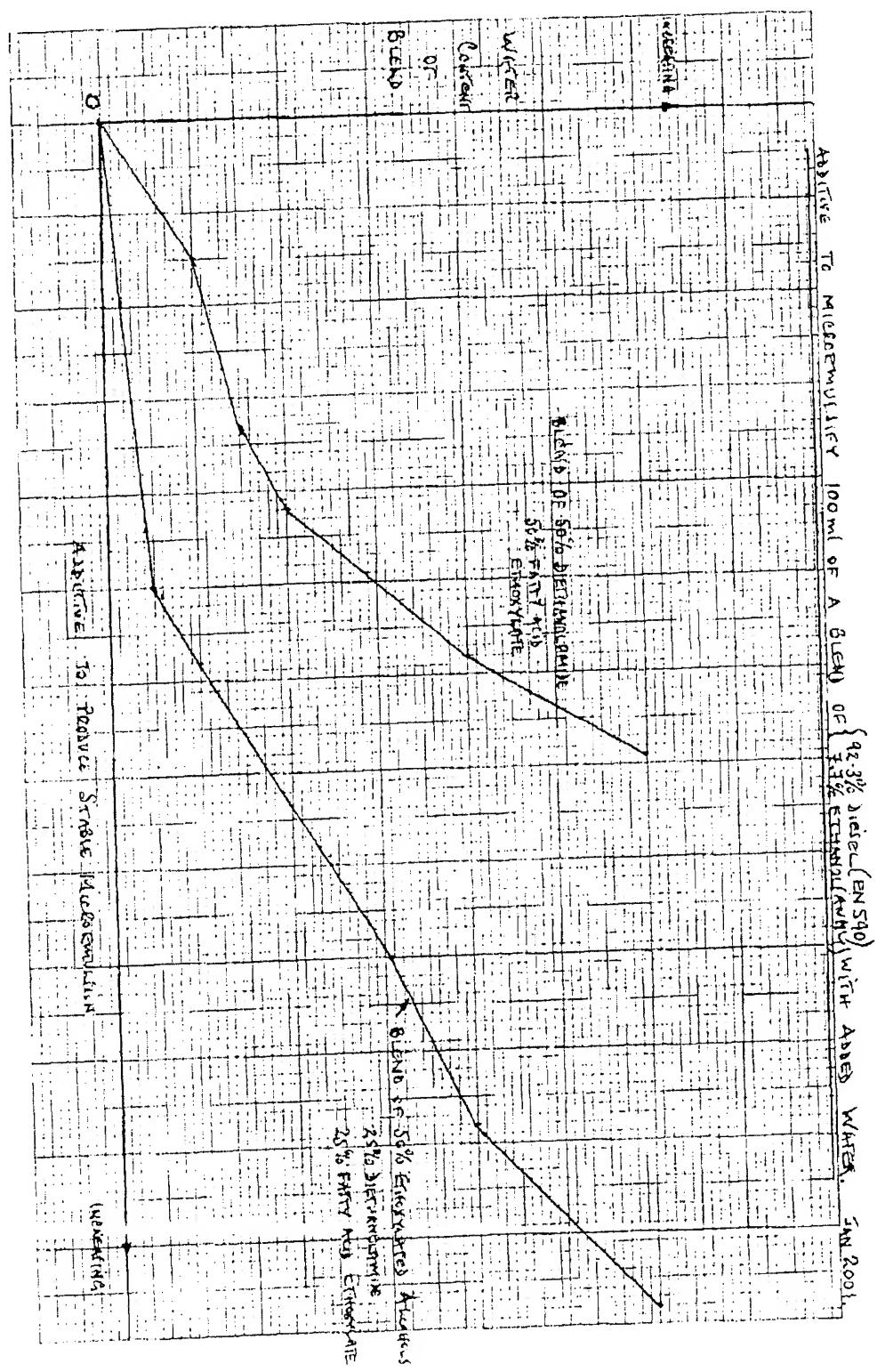
43. A method of running an internal combustion engine comprising the use of a fuel according to claim 13.

40 44. A method according to claim 43 **characterised in that** the toxicity of the exhaust emissions from the engine are reduced.

45. A fuel additive or a fuel composition substantially as described with reference to the accompanying examples.

50

55





European Patent  
Office

## EUROPEAN SEARCH REPORT

Application Number  
EP 01 10 1960

DOCUMENTS CONSIDERED TO BE RELEVANT			CLASSIFICATION OF THE APPLICATION (Int.Cl.7)
Category	Citation of document with indication, where appropriate, of relevant passages	Relevant to claim	
D, X	GB 2 217 229 A (UNIV CITY ;ENERSOLVE CHEMICAL COMPANY LIM (GB)) 25 October 1989 (1989-10-25)	1, 3-7, 11, 13-26, 41-43 28-30	C10L1/14 C10L1/10 C10L1/02 C10L1/32 C10L10/02
Y	* page 5 - page 6 *	---	
Y	WO 00 36055 A (WILLIAMSON IAN VERNON ;AAE HOLDINGS PLC (GB); HAZEL CLIFFORD JAMES) 22 June 2000 (2000-06-22) * the whole document *	28-30	
D, A	WO 98 17745 A (WILLIAMSON IAN VERNON ;HAZEL CLIFFORD JAMES (GB)) 30 April 1998 (1998-04-30) * the whole document *	1-45	
D, A	WO 99 35215 A (WENZEL DEBORAH) 15 July 1999 (1999-07-15)	19, 20, 23-25, 28, 29, 31-36	
	* the whole document *	---	TECHNICAL FIELDS SEARCHED (Int.Cl.7)
A	WO 00 31216 A (PURE ENERGY CORP) 2 June 2000 (2000-06-02) * claim 1 *	37-40	C10L
A	WO 99 20715 A (PURE ENERGY CORP) 29 April 1999 (1999-04-29) * the whole document *	14, 21, 22	
A	WO 99 52996 A (THORLEY DAVID ;STREET PETER (GB); COVAL TECHNOLOGIES LIMITED (GB)) 21 October 1999 (1999-10-21)		
A	EP 0 957 152 A (KAO CORP) 17 November 1999 (1999-11-17)		
		-/-	
The present search report has been drawn up for all claims			
Place of search	Information portion of the search	Examiner	
THE HAGUE	13 July 2001	De La Morinerie. 3	
CATEGORY OF CITED DOCUMENTS			
X	particularly relevant taken alone	1 theory or principle underlying the invention	
Y	particularly relevant if combined with another document of the same category	1 earlier patent document, but published on, or after the filing date	
A	technological background	1) document cited in the application	
O	non-written disclosure	1) document cited for other reasons	
P	intermediate document	2) member of the same patent family - corresponding document	



European Patent  
Office

## EUROPEAN SEARCH REPORT

Application Number

EP 01 10 1960

DOCUMENTS CONSIDERED TO BE RELEVANT			CLASSIFICATION OF THE APPLICATION (Int.Cl.7)												
Category	Citation of document with indication, where appropriate, of relevant passages	Relevant to claim													
A	WO 99 44732 A (KELLY ELIZABETH MARY TERESA ;CLOHESSY JUSTIN PETER (ZA); LUNDIN IN) 10 September 1999 (1999-09-10) ---														
A	US 4 110 283 A (CAPELLE ANTHONY) 29 August 1978 (1978-08-29) ---														
A	DE 29 40 782 A (HENKEL KGAA) 16 April 1981 (1981-04-16) -----														
			TECHNICAL FIELDS SEARCHED (Int.Cl.7)												
<p>The present search report has been drawn up for all claims</p> <table border="1" style="width: 100%; border-collapse: collapse;"> <tr> <td style="width: 33%;">Place of search</td> <td style="width: 33%;">Date of completion of the search</td> <td style="width: 33%;">Examiner</td> </tr> <tr> <td>THE HAGUE</td> <td>18 July 2001</td> <td>De La Morinerie, B</td> </tr> </table> <p>CATEGORY OF CITED DOCUMENTS</p> <table border="0" style="width: 100%; border-collapse: collapse;"> <tr> <td style="width: 33%;">X particularly relevant if taken alone</td> <td style="width: 33%;">Y particularly relevant if combined with another document of the same category</td> <td style="width: 33%;">A technological background</td> </tr> <tr> <td>Z non-written disclosure</td> <td>Q intermediate document</td> <td></td> </tr> </table> <p style="text-align: center;">1. theory or principle underlying the invention 2. earlier patent document, but published on, or after the filing date 3. document cited in the application 4. document cited for other reasons 5. member of the same patent family, corresponding document</p>				Place of search	Date of completion of the search	Examiner	THE HAGUE	18 July 2001	De La Morinerie, B	X particularly relevant if taken alone	Y particularly relevant if combined with another document of the same category	A technological background	Z non-written disclosure	Q intermediate document	
Place of search	Date of completion of the search	Examiner													
THE HAGUE	18 July 2001	De La Morinerie, B													
X particularly relevant if taken alone	Y particularly relevant if combined with another document of the same category	A technological background													
Z non-written disclosure	Q intermediate document														

ANNEX TO THE EUROPEAN SEARCH REPORT  
ON EUROPEAN PATENT APPLICATION NO.

EP 01 10 1960

This annex lists the patent family members relating to the patent documents cited in the above mentioned European search report.  
 The members are as contained in the European Patent Office EDP file on  
 The European Patent Office is in no way liable for these particulars which are merely given for the purpose of information.

18-07-2001

Patent document cited in search report		Publication date	Patent family member(s)		Publication date
GB 2217229	A	25-10-1989	NONE		
WO 0036055	A	22-06-2000	AU	1788500 A	03-07-2000
WO 9817745	A	30-04-1998	AU	4710097 A	15-05-1998
			BR	9711430 A	31-10-2000
			DE	19782068 T	11-11-1999
			GB	2334964 A, B	08-09-1999
			JP	2001502374 T	20-02-2001
			SE	9901364 A	16-04-1999
WO 9935215	A	15-07-1999	AU	2316299 A	26-07-1999
			EP	1047756 A	02-11-2000
WO 0031216	A	02-06-2000	US	6017369 A	25-01-2000
			AU	1741100 A	13-06-2000
			FI	20011065 A	21-05-2001
			US	6190427 B	20-02-2001
			US	2001003881 A	21-06-2001
WO 9920715	A	29-04-1999	US	6074445 A	13-06-2000
			AU	731702 B	05-04-2001
			AU	9809798 A	10-05-1999
			EP	1027410 A	16-08-2000
			NO	20001972 A	08-06-2000
			US	6183524 B	06-02-2001
			US	2001005956 A	05-07-2001
			ZA	9809525 A	20-04-1999
WO 9952996	A	21-10-1999	GB	2336119 A	13-10-1999
			AU	3430499 A	01-11-1999
EP 0957152	A	17-11-1999	WO	9727271 A	31-07-1997
WO 9944732	A	10-09-1999	AU	2941799 A	20-09-1999
US 4110283	A	29-08-1978	NONE		
DE 2940782	A	16-04-1981	NONE		

For more details about this annex : see Official Journal of the European Patent Office, No. 12/82